**Displacement reactions (Level 1) examiners tips: Read these please!**

• a more reactive metal displaces another less reactive metal from a solution if its salt

• be sure to refer to the Activity Series provided to you in the Resource booklet

• link all colours to their species

• refer to initial and final colours

• double check that your equations are balanced

*to achieve with Merit you must explain displacement reactions in terms of both atoms and ions*

**Also…”don’t be daft”**

remember that COLOURLESS solutions are just that! never clear!

more reactive elements are higher on the Activity Series (or to the left) in your Resource booklet

iron metal is made up of ATOMS with the symbol Fe

Fe(II) or Fe(III) are IONS

similarly copper METAL is Cu

Cu(II) is the symbol for copper ions

the symbol for potassium is K, not P!!!! (K comes from the medieval Latin word Kalium)

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